

Léon Dupont,^{a*} Pascal De Tullio,^b Stéphane Boverie^b and Bernard Pirotte^b^aUnité de Cristallographie, Institut de Physique – Bât. B5, Université de Liège, Allée du 6 Août, 17, B-4000 Liège, Belgium, and ^bService de Chimie Pharmaceutique, Institut de Pharmacie – Bât. B36, Université de Liège, Avenue de l'Hôpital, 1, B-4000 Liège, Belgium

Correspondence e-mail: leon.dupont@ulg.ac.be

Key indicators

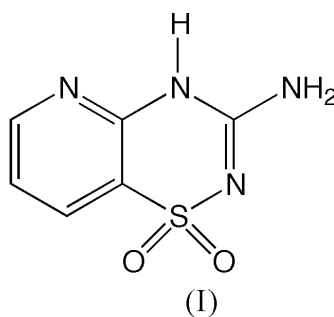
Single-crystal X-ray study
 $T = 293\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.004\text{ \AA}$
 R factor = 0.032
 wR factor = 0.090
Data-to-parameter ratio = 8.4For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.3-Amino-4*H*-pyrido[2,3-*e*]-1,2,4-thiadiazine 1,1-dioxide

The title compound, $\text{C}_6\text{H}_6\text{N}_4\text{O}_2\text{S}$, was prepared for structural comparison with diazoxide (7-chloro-3-methyl-4*H*-1,2,4-benzothiadiazine 1,1-dioxide) and other 3-alkylaminopyrido-thiadiazine 1,1-dioxides known to be potassium channel openers. Particular attention was paid to the tautomeric conformation adopted by the compound in the crystalline state, which is found to be the 4*H*-form.

Received 17 May 2001
Accepted 14 June 2001
Online 22 June 2001

Comment

Although most of the published X-ray data on pyridothiadiazine dioxides are related to compounds with the pyridinic nitrogen in the 7-position of the pyridothiadiazine ring, this example, (I), of a compound with the N atom in the 5-position will complete our information on the influence of this position on the tautomeric behaviour of pyridothiadiazine dioxides. It confirms the predominance of the 4*H*-tautomeric form. These results may help to establish in such molecules which are the important chemical and geometrical parameters required (pharmacophore) for their biological activity.



Experimental

The title compound was synthesized at the Laboratory of Medicinal Chemistry of Liège, according to the method of Kotovskaya *et al.* (1979). Crystals were obtained by slow evaporation of a methanol solution.

Crystal data

 $\text{C}_6\text{H}_6\text{N}_4\text{O}_2\text{S}$
 $M_r = 198.21$
Orthorhombic, *Pbca*
 $a = 9.7545(6)\text{ \AA}$
 $b = 12.0427(9)\text{ \AA}$
 $c = 13.5242(18)\text{ \AA}$
 $V = 1588.7(3)\text{ \AA}^3$
 $Z = 8$
 $D_x = 1.657\text{ Mg m}^{-3}$ Cu $K\alpha$ radiation
Cell parameters from 34 reflections
 $\theta = 24.7\text{--}40.7^\circ$
 $\mu = 3.43\text{ mm}^{-1}$
 $T = 293(2)\text{ K}$
Prism, colourless
 $0.34 \times 0.23 \times 0.11\text{ mm}$

Data collection

Stoe–Siemens AED four-circle diffractometer

 ω scansAbsorption correction: ψ -scan (EMPIR; Stoe & Cie, 1987) $T_{\min} = 0.388$, $T_{\max} = 0.704$

1077 measured reflections

1077 independent reflections

829 reflections with $I > 2\sigma(I)$ $\theta_{\max} = 57.4^\circ$ $h = 0 \rightarrow 10$ $k = 0 \rightarrow 13$ $l = 0 \rightarrow 14$

2 standard reflections

frequency: 60 min

intensity decay: 3%

RefinementRefinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.032$ $wR(F^2) = 0.090$ $S = 0.96$

1077 reflections

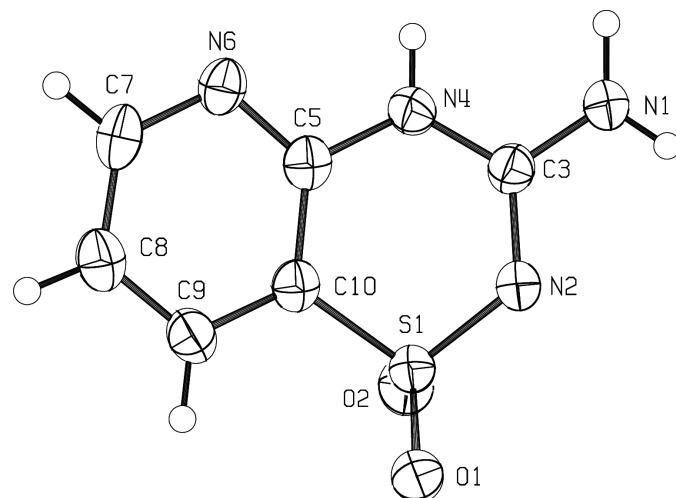
128 parameters

H atoms treated by a mixture of independent and constrained refinement

 $w = 1/[\sigma^2(F_o^2) + (0.0560P)^2 + 0.7861P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} < 0.001$ $\Delta\rho_{\max} = 0.26 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{\min} = -0.23 \text{ e } \text{\AA}^{-3}$

Extinction correction: SHELXL97

Extinction coefficient: 0.0043 (4)

**Figure 1**

The molecular structure of the title compound with the atom-labelling scheme. Displacement ellipsoids are shown at the 50% probability level and H atoms are shown as small circles or arbitrary radii.

Table 1Hydrogen-bonding geometry (\AA , $^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$N4-H4 \cdots N2^i$	0.84 (3)	2.22 (3)	3.047 (3)	172 (3)
$N11-H112 \cdots O1^i$	0.92 (3)	1.96 (3)	2.868 (3)	173 (3)
$N11-H111 \cdots N6^{ii}$	0.86 (3)	2.13 (3)	2.997 (3)	177 (3)

Symmetry codes: (i) $\frac{1}{2} + x, \frac{1}{2} - y, -z$; (ii) $x - \frac{1}{2}, \frac{1}{2} - y, -z$.

H atoms were constrained (included as riding atoms) except those on N4 and N11, which were refined, with isotropic displacement parameters fixed at $1.2U_{eq}$ of the parent atom.

Data collection: DIF4 (Stoe & Cie, 1987); cell refinement: DIF4; data reduction: REDU4 (Stoe & Cie, 1987); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEPIII (Burnett & Johnson, 1996); software used to prepare material for publication: SHELXL97.

The authors thank M. M. Vermeire for his helpful assistance in the diffractometry measurements and the Belgian FNRS (Fonds National de la Recherche Scientifique) for financial support.

References

- Burnett, M. N. & Johnson, C. K. (1996). ORTEPIII. Report ORNL-6895. Oak Ridge National Laboratory, Tennessee, USA.
- Kotovskaya, S. K., Mokrushina, G. A., Potovskii, I. Y., Pidemskii, E. L., Goleneva, A. F. & Vysokova, T. Y. (1979). *Khim. Farm. Zh.* **13**, 54–57.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Stoe & Cie (1987). DIF4, REDU4 and EMPIR. Version 6.2. Stoe & Cie, Darmstadt, Germany.